Kurt Wüthrich, the ETH Zürich, and the Development of NMR Spectroscopy for the Investigation of Structure, Dynamics, and Folding of Proteins

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General George Patton was the military leader of the American Third Army in World War II. Starting in 1944, the Third Army advanced through Brittany, round Paris, up the Marne and the Moselle, across the Rhine, and eventually into Czechoslovakia in April 1945. Patton was known for his short and sharp addresses to his troops. In one of those speeches, he welcomed the 761st Black Panther Tank battalion, the first battalion to admit black soldiers to the army. He ended his speech for the soldiers with the following sentences:

™Everyone has their eyes on you and [is] expecting great things from you. Most of all, your race is looking forward to your success. Don't let them down, and damn you, don't let me down! If you want me you can always find me in the lead tank.^{"[1]}

Nuclear magnetic resonance (NMR) spectroscopy is one of the most powerful techniques to study the structure, dynamics, and folding of proteins. This view has been recognized by the Nobel prize committee by awarding one half of the 2002 Nobel prize in Chemistry to Prof. Kurt Wüthrich, of the Eidgenössische Technische Hochschule (ETH) Zürich, "for his development of nuclear magnetic resonance spectroscopy for determining the threedimensional structure of biological macromolecules in solution.∫

And indeed, when summarizing Wüthrich's scientific contributions to this field, one has to tell the story of the development of NMR spectroscopy for the study of biological macromolecules in general and of proteins in particular (Table 1). This development has been driven by a battalion of outstanding researchers, amongst which, like General Patton during the Allied invasion in 1944, Wüthrich could always be found in the lead tank. And similarly, Wüthrich's scientific orientation has been and continues to be focused, ambitious, and straight.

Wüthrich's first encounter with magnetic resonance was during his PhD research with Silvio Fallab in Basel, where he studied the mechanism of Cu²⁺-catalyzed reactions^[2] and VO²⁺ complexes^[27] by using electron spin resonance spectroscopy. He joined Robert E. Connick as a postdoctoral fellow at Berkeley, California, where he published his first NMR spectroscopy paper entitled ™Nuclear magnetic resonance relaxation of oxygen-17 in aqueous solutions of vanadyl perchlorate and the rate of elimination of water molecules from the first coordination sphere". The paper deals with ¹⁷O NMR spectroscopy of paramagnetic vanadylate perchlorate.^[28] Interestingly, Wüthrich later studied the dynamics of hydration in proteins–old questions investigated under a new angle.

From paramagnetic inorganic salts to paramagnetic proteins

The first NMR spectrum of the protein ribonuclease was studied by Saunders et al.^[29] in 1957. Wüthrich's first protein NMR studies, together with Bob Shulman, dealt with the paramagnetic protein cyanometmyoglobin and were published in 1968.[3] In 1970, the young Privatdozent Wüthrich published an article in Chimia on the ™study of the spatial structure of protein molecules by NMR spectroscopy".[4] This was seven years after Kendrew and Perutz had solved the first three-dimensional structures of the proteins myoglobin^[30] and haemoglobin^[31] by X-ray crystallography. By showing whether NMR investigations of proteins in solution and X-ray studies of proteins in their crystalline form would yield similar spatial structures of proteins under the widely different experimental conditions, Wüthrich intended to provide an important cross-validation for both techniques. In addition, he continued, NMR spectroscopy could be applied to investigate the structural properties of proteins under conditions that do not yield crystals.

What in hindsight turned out to be the topic of his life must have evoked some scepticism at the time of this initial publication. It took Wüthrich's courage and determination for over thirty years to prove that his claim to be able ™to solve the structure of a protein["], put forward in his unique and charming

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English with a Swiss accent on behalf of the whole NMR community, could indeed be fulfilled: NMR spectroscopy can yield structures of proteins.[32] In fact, NMR spectroscopy goes far beyond a purely static description of proteins; it is able to describe complete studies of the dynamics of a polypeptide chain from its random coil state to intermediates populated during folding, to the different manifestations of proteins from soluble monomeric forms, to insoluble fibrillic aggregates. And ultimately, the size barriers that restricted NMR spectroscopy to comparatively small systems in the past may finally be pushed back with the TROSY technology that was introduced by the Wüthrich group in 1997.^[25]

"If you want to solve the structure of a protein... $^{\prime\prime[78]}$

Protein structure determination by NMR spectroscopy involves two steps. Each atom in the molecule generates a signal in the NMR spectrum. Even in a small protein such as ubiquitin whose one-dimensional spectrum is shown in Figure 1 A, there are more than a thousand hydrogen atoms, all of which need to be assigned. The resonance assignment then forms the basis for the extraction of conformational dependent parameters such as through-space-mediated NOEs and through-bond-mediated scalar J couplings that can be used to obtain distance and torsion angle restraints.

1. NMR Resonance Assignment Procedure: 2D FT-NMR Spectroscopy and the COSY - NOESY Sequential Assignment Approach

With a thousand hydrogen atoms, it is impossible to resolve all resonance lines in a one-dimensional spectrum. It was clear to Wüthrich early on that the two-dimensional^[33] methodologies, such as the COSY^[34] and NOESY^[35, 36] experiments developed by his colleague Richard R. Ernst^[37] in the Department of Physical Chemistry at the ETH Zürich, would have to be transferred from small molecules and their applicability proven for large proteins. At the time, the research situation at the ETH of two groups with complimentary focus, each pushing the limits of the novel NMR technology, was enormously successful. In 1980, the groups of Wüthrich and Ernst published the first application of the novel two-dimensional experiments for the study of proteins.^[12, 13] From an NMR point of view, the hydrogen atoms of amino acids in polypeptides form spin systems of coupled protons that resonate at a unique and characteristic chemical shift. Spins two or three bonds apart can be correlated through scalar $2J$ and $3J$ coupling constants in COSY experiments (Figure 1 B). Such correlation leads to the assignment of resonances within a given amino acid. Connection of individual residues can be obtained from sequential, through-space-mediated transfers in NOESY experiments that reveal cross peaks between protons that are closer than 5 Å together (Figure 1C).

Figure 1. A) One-dimensional NMR spectrum, B) two-dimensional COSY spectrum, C) twodimensional NOESY spectrum of the protein ubiquitin in H₂O at 600 MHz.

2. Protein Structure Calculation

On the basis of resonance assignment and the analysis of NOE patterns, Wüthrich first turned to the question of defining the conformational dependence of NMR parameters such as chemical shifts,^[10] short and medium-range NOEs,^[14b] and scalar J coupling constants^[17] by comparing the two extremes of states that a protein can adopt (the native folded state and the random coil state of a protein), using model peptides GGXGG with assumed random-coil conformational averaging.

In 1979, it could be shown that distance geometry algorithms can yield structures of proteins,[38] and in 1985, the laboratories of Wüthrich^[18a] and of Kaptein^[39] solved the first experimental NMR spectroscopy protein structures, which were soon followed by the first NMR spectroscopy structure of a DNA by Hare and Reid.^[40] The distance geometry method may be compared with a neatly woven three-dimensional spider's web. In the interior of a protein, each proton is surrounded by a large number of protons less than 5 Å apart, just as a knot in a spider's web is connected through short strings to the next layer of the web with corresponding knots. Each knot in the second sphere is the center of a new network of connectivity, and the uniform distribution of distances in globular proteins allows for the

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calculation of protein structure based on a large number of short and locally defined NOEs. This procedure is the essence of the distance geometry algorithm. In this algorithm, the connections between the knots in the web are the experimental restraints (such as NOE-derived distances and torsion-angle restraints from scalar coupling constants) which are taken together with the covalent structure of the protein to create a matrix of interatomic distances between all atoms. Starting from random coordinates that only fulfill the covalent constraints, the experimental restraints are embedded.

In 1988, Wüthrich and his colleague Robert Huber, an X-ray crystallographer from the Max Planck Institute, Martinsried, used tendamistat, an α -amylase inhibitor whose structure was first solved independently,^[41, 18c] as a model to compare the results of the two independent structure determinations by NMR spectroscopy and X-ray crystallography (Figure 2).[21b] The result was overwhelmingly positive for both techniques: In the interior of the protein the two structures were nearly identical, while only on the protein surface could a small number of local differences be identified. For most residues on the surface of the protein, the solution structure ™appeared more disordered than the crystal structure, with the exception of tyrosine-14, which was not observed in the X-ray diffraction.["] In a subsequent publication, Huber and Wüthrich also showed that the NMR spectroscopy solution structure could be used as model to solve the X-ray crystallographic phases with similar quality to isomorphous replacement techniques.[21c]

Figure 2. A) Family of structures of tendamistat, an α -amylase inhibitor determined by NMR spectroscopy, B) ribbon diagram of the structure with lowest energy.

3. Protein Dynamics and Beyond

It is a particular strength of NMR spectroscopy to investigate more than the static three-dimensional structure of proteins. Dynamic applications range from the internal dynamics of individual atomic groups, to interactions of the protein with solvent molecules such as water or chemical denaturants, to the process of protein folding. One early example from the Wüthrich group is the determination of the free activation energies of aromatic ring flips in bovine pancreatic trypsin inhibitor $(BPTI).$ ^[42, 6] The systematic investigation of hydrogen-exchange dynamics as a function of pH value, temperature, and concentration of denaturants^[11, 19] laid the ground work for the investigation of the kinetics of protein folding itself with atomic resolution by Roder and Wüthrich,^[20] a highly influential experiment that at the time may not have received its due recognition. Lastly, NMR spectroscopy can be used to map out the hydration sphere of proteins^[22] and oligonucleotides^[43] and to provide bounds for residence life times of individual water molecules.

4. The Protein Structure Gallery

NMR spectroscopy has made a major contribution to the determination of structures of proteins and oligonucleotides: 20% of the approximately 14 000 structures deposited in the protein data bank (PDB) to date have been solved by NMR spectroscopy. Over the years, Wüthrich's group has contributed with more than 50 structures (summarized in Table 2 and

Figure 3). The three-dimensional structures of biological macromolecules are fundamental for our understanding of cell biology. Again, NMR spectroscopy provides important insight beyond the description of only the conformationally rigid part of proteins: the structure of the benign cellular form of the prion protein^[62, 70-72] solved by Wüthrich's group shows that half of the prion protein backbone is well-ordered whereas the other half is a highly mobile, extended coil.

5. NMR Spectroscopy for Large Proteins

For a long time, NMR spectroscopy could only solve structures of relatively small proteins. In 1997, Pervushin, Wüthrich, and coworkers^[25] published a communication in which they showed that for amide sites in a very large protein, in which all hydrogen atoms but the exchangeable ones such as the amide hydrogen have been replaced with deuterium, the effect of crosscorrelated relaxation of different relaxation mechanisms such as NH dipole - dipole relaxation and N chemical shift anisotropy^[76] leads to enhanced relaxation for the downfield component of the H^N doublet, while for the upfield component, the two relaxation processes can mutually cancel. Since the chemical shift anisotropy is field dependent and the dipole-dipole relaxation is field independent for large molecules, the two

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Figure 3. Gallery of selected protein structures solved by the Wüthrich group: A) Proteinase inhibitor IIA from bull seminal plasma (PDB accession code: 1BUS),^[18a] B) pulmonary surfactant-associated polypeptide SP-C (1SPF),^[59] C) bovine pancreatic trypsin inhibitor (1PIT),^[52] D) the complex of cyclosporin A bound to cyclophilin A $(3CYS),^[61]$ E) the periplasmic chaperone FimC (1BF8),^[67] G) the structured part of the human prion protein (1QLX).^[72]

effects will mutually cancel at a magnetic field of approximately 950MHz (Figure 4). It was again the Wüthrich group who clearly recognized the importance of this observation and its tremendous impact for the NMR spectroscopy of larger protein complexes–with the latest record documented in a study of the molecular chaperone complex GroEL - GroES with a molecular weight of 900 K.[77]

Final Remarks

The development of NMR spectroscopy as one of the two major tools for the study of structure, dynamics, and folding of proteins and their interaction with small and large molecules from drugs to molecular chaperones has been and continues to be intimately linked with the research of Kurt Wüthrich. His school is now spread throughout the world and is continuing to advance this exciting field, unique and interdisciplinary with applications and influences to and from physics, chemistry, biology, and medicine.

Figure 4. The TROSY effect.^[25] In a ¹⁵N-labeled protein, the amide resonance line is split into a doublet due to the ¹J(N,H) coupling constant of approximately 90 Hz. Due to crosscorrelated relaxation of the dipole - dipole relaxation and the chemical shift anisotropy (CSA) relaxation mechanism, the two components of the doublet have different line width (blue line).The CSA contribution is field dependent and the condition at which destructive interference of the two relaxation mechanisms leads to mutual cancellation is given.

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- [78] Note added in proof: Kurt Wüthrich uses his belt to explain the process of NMR structure determination, as can be seen in the video recording of his lecture given on December 8, 2002, at the Magna Aula, Stockholm University, Sweden; www.nobel.se.

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